# Convenient one-pot method for the preparation of polysubstituted benzo[b]- and naphtho[1,2-b]-furans and -thiophenes 

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#### Abstract

$N$-(Phenoxymethyl)- and $N$-(phenylthiomethyl)-benzotriazoles are versatile substrates for the preparation of benzofurans and benzothiophenes by insertion reactions of their anions into alkyl and aryl aldehydes and in situ cyclization of the $\alpha$-aryloxy and $\alpha$-arylthio ketones thus formed. $N$-(Naphthyloxy)- and $N$-(naphthylthio)-benzotriazoles are similarly efficient precursors for naphthofurans and naphthothiophenes.


## Introduction

The constant and growing interest in the development of new efficient and general synthetic methods for the preparation of fused heterocyclic systems involving furan and thiophene subunits is justified by their well-established valuable physiological and pharmacological properties. ${ }^{1-4}$ In addition, recent technical applications of polysubstituted benzofurans and benzothiophenes, including numerous fluorescent dyes used as retrograde tracers, and $\mathrm{Ca}^{2+}$ and $\mathrm{Mg}^{2+}$ fluorescent indicator conjugates, etc. ${ }^{4,5}$ increase their significance.
Benzo[b]furans $\mathbf{A}$ and benzo[ $b]$ thiophenes $\mathbf{C}$ (Scheme 1) are

each available by two general routes: (i) the construction of a fused benzene ring starting from 2 -substituted furans ${ }^{6 a}$ or thiophenes, ${ }^{6 b-d}$ and (ii) the formation of a furan ${ }^{6 e-l}$ or thiophene ${ }^{61-n}$ moiety beginning with $\mathrm{S}(\mathrm{O})$-substituted derivatives of (thio)phenols. The extensive literature devoted to such methods has been summarized. ${ }^{1,7-9}$

Among the known methods, both benzo[b]furans $\mathbf{A}$ and benzo $[b]$ thiophenes $\mathbf{C}$ have most frequently been obtained by acid-catalyzed cyclizations of $\alpha$-substituted aryl(thio)oxycarbonyl compounds of general type $\mathbf{B},{ }^{1,9}$ a route which has proven to be of great value for the synthesis of a wide variety of $\mathbf{A}$ and C. The assembling of a benzene ring from derivatized furans or thiophenes has been of lesser synthetic importance (annulation of a benzene ring; ${ }^{2}$ the scope and limitations of this method for the benzothiophene series were recently given ${ }^{6 d}$ ). Aryl(thio)oxy keto-derivatives of type $\mathbf{B}$ are often prepared by means of condensation of (thio) phenols and $\alpha$-halogeno- $\beta$-keto esters in the presence of bases (for an example, see ref. 10). The analogous $\alpha$-aryl(thio)oxy aldehydes are frequently used in a protected (acetal) form, and are prepared from (thio)phenols and $\alpha$-halo acetals. ${ }^{11-13}$ Known methods for the preparation of both benzo$[b]$ furans and benzo[b]thiophenes are briefly compared later in this paper, but no existing method is completely general. The preparations of aryl(thio)oxy ketones of type $\mathbf{B}$ also possess certain limitations, especially as far as varieties of $R^{1}$ and $R^{2}$ substituent are concerned. Hence, novel general routes for both benzo[b]furans and benzo[b]thiophenes retain synthetic importance.

We recently developed an efficient method for carbon insertion, which allowed the preparation of several new $\alpha$-aryl(thio) oxy ketones in moderate to good yields (53-86\%). ${ }^{14} \mathrm{We}$
have now extended this methodology to provide convenient one-pot procedures to prepare polysubstituted benzo[ $b]$ furans and benzo[ $b]$ thiophenes.

## Results and discussion

1-(Aryloxymethyl)benzotriazoles 3a-c were prepared by reactions of 1 -chloromethylbenzotriazole 1 with polysubstituted phenols 2a-c (Scheme 2). Lithiation of phenyl ethers 3a-c with $\mathrm{Bu}^{n} \mathrm{Li}$, followed by quenching with an appropriate electrophile, afforded the substituted benzotriazole derivatives $\mathbf{4 a}-\mathbf{f}$. Alkyl intermediates 4 could be isolated, and this was done for the cases of $\mathbf{4 a}$ and $\mathbf{4 b}$. In general, $\mathbf{4 a}-\mathbf{f}$ were used directly in the next step. Treatment of $\mathbf{4 a - f}$ with an additional equivalent of $\mathrm{Bu}^{n} \mathrm{Li}$, and subsequently with an aldehyde formed intermediates 5a-f. Once again, derivatives $\mathbf{5}$ could be isolated and characterized, and this was done in the case of $\mathbf{5 b}$. However, in general, 5a-f were treated directly without isolation with $\mathrm{ZnBr}_{2}$ (neat mixture of $140^{\circ} \mathrm{C}$ ), to afford the corresponding $\alpha$-aryloxy ketones $\mathbf{6 a - f}$ as a result of pinacol-type rearrangement (cf. ref. 14). Here also, 6a-f were generally not isolated; the resulting mixture was heated at $175-180^{\circ} \mathrm{C}$ for $10-24 \mathrm{~h}$, to give the desired polysubstituted benzo[b]furans $7 \mathbf{7}-\mathbf{f}$ in good overall yields ( $49-80 \%$ for four steps). Several $\alpha$-aryloxy ketones $\mathbf{6 a - e}$ were isolated in the course of the reactions in order to prove the reaction sequence; on further treatment with $\mathrm{ZnBr}_{2}$ at elevated temperature, $\mathbf{6 a - e}$ each gave the corresponding benzofuran $7 \mathrm{a}-\mathrm{e}$, respectively.
Following a similar protocol, various substituted benzo[b]thiophenes 7 g and 7 h (Scheme 2) were prepared in one-pot procedures from 1-(arylthiomethyl)benzotriazoles $\mathbf{3 d}$ and $\mathbf{3 e}$ via the corresponding alkylated analogs $\mathbf{4}$, and arylthio ketones 6. Just as for the sequence leading to the analogous benzofurans, intermediates ( $\mathbf{4 h}$ and $\mathbf{6 h}$ ) were isolated during the course of these similar transformations in the benzo $[b]$ thiophene series. Cyclization of ketones $\mathbf{6 g}$ and $\mathbf{6 h}$ occurred smoothly in hot $\left(120-125^{\circ} \mathrm{C}\right)$ polyphosphoric acid, leading to the expected benzo $[b]$ thiophenes 7 g and 7 h in good overall yields $(75-78 \%)$. Polyphosphoric acid is the reagent/solvent of choice for the benzo[b]thiophene series: when the ketone $\mathbf{6 h}$ was treated with zinc bromide at elevated temperature under conditions similar to the preparation of the benzo[b]furan series, only a low yield ( $c a .5 \%$, according to the GC-MS data) of the desired product 7 h was obtained after 40 h of heating; instead, di-p-tolyl disulfide ( $15 \%$ ) and 1,4 -di- $p$-toluoylbutane ( $35 \%$ ) were isolated from the reaction mixture by column chromatography together with unreacted starting ketone 6h. This may be explained by the thermal instability of ketones $\mathbf{6 g}$, $\mathbf{6 h}$ derived from thiophenol: the structures of the byproducts clearly indicate a radical cleavage of the tert-C-S bond. The linear structure of 1,4-di-p-toluoylbutane was formed

via migration of the unpaired electron to the methyl carbon atom followed by the recombination of two radical species.

Cyclization of ketones of general type 6 (Scheme 3) is considered to be the best way to prepare symmetrical 2,3dialkylsubstituted benzo[b]furans and benzo $[b]$ thiophenes. ${ }^{1}$ When the corresponding unsymmetrically substituted ketones $\left(\mathrm{R}^{1} \neq \mathrm{R}^{2} \neq \mathrm{H}\right)$ are employed, mixtures of regioisomers with substituents partly exchanged at the $C(2)$ and $C(3)$ positions were usually obtained. ${ }^{1,15}$ The isomer ratio has been shown to depend greatly on the reaction temperature in the benzofuran series, and, accordingly, in order to obtain benzofurans or benzothiophenes free from the corresponding regioisomers, special reagents/methods need to be used. ${ }^{1}$ In our reactions involving the use of both zinc bromide and polyphosphoric acid, the formation of alternative isomers was largely suppressed, leading to predominant isolation of products $7 \mathbf{a}-\mathbf{h}$ with the expected unrearranged structure. However, in the ${ }^{1} \mathrm{H}$ NMR spectra of some products, i.e. 7a, a second weak set of signals for the corresponding positional isomer was observed. The GC-MS spectra of initially isolated $\mathbf{7 a}$ and $\mathbf{7 b}$ also show existence of a second peak of the same molecular weight, but with different fragmentation. The ratio of 'regular' 7 and 'reversed' 7' products, according to the GC-MS data, was $c a .13: 1$ in the case of 7a and $20: 1$ in the case of $\mathbf{7 b}$. Cyclodehydration reactions leading to the formation of benzofurans and benzothiophenes with substituents swapped at the $C(2)$ and $C(3)$ positions, probably, involve cations of type 8, which exist in equilibrium with the corresponding 3-membered heterocyclic cations 9: depending



6


7
$7^{\prime}$

Scheme 3
on the lability of the $\mathrm{X}-\mathrm{C}-\mathrm{R}^{2}$ and $\mathrm{X}-\mathrm{C}-\mathrm{R}^{1}$ bonds, either $\mathbf{1 0}$, or $\mathbf{8}$ can be formed, and thus variously substituted benzofurans or benzothiophenes can result (Scheme 3).

Naphthofurans 16a and 16b and naphthothiophenes 16c and $16 \mathbf{d}$ with various substituents at the $C(2)$ and $C(3)$ positions were prepared (Scheme 4) in overall yields of $71-83 \%$, by a route similar to that for the benzo-furan and -thiophene series described above. The (benzotriazol-1-yl)methyl 1-naphthyl (thio)ethers 12a and 12b were isolated ( $95 \%$ yield in both cases), but all the subsequent transformations were performed in onepot sequences: lithiation and addition of alkyl halide afforded substituted (thio)naphthols 13a-d, which were treated in situ

with more $\mathrm{Bu}^{n} \mathrm{Li}$ and various aldehydes to form the appropriate alcohols 14a-d. The Lewis acid $\left(\mathrm{ZnBr}_{2}\right)$ induced carbon insertion and formed the corresponding ketones 15a-d. For the preparation of naphthofuran 16a the temperature was raised to $175-180^{\circ} \mathrm{C}$, while for the preparation of $\mathbf{1 6 b}$ and naphthothiophenes $\mathbf{1 6 c}$ and $\mathbf{1 6 d}$ polyphosphoric acid was added, followed by an increase in temperature to $120-125^{\circ} \mathrm{C}$. The high temperature (neat) cyclization of the intermediate $\mathbf{1 5 b}$ gave a mixture of desired benzophenone $\mathbf{1 6 b}$ and a by-product whose structure appears to include a naphthopyran ring (according to NMR spectra of the crude reaction mixture). However, all attempts to separate $\mathbf{1 6 b}$ and the by-product failed. Therefore we used PPAcatalyzed cyclization for the preparation of $\mathbf{1 6 b}$. The reactions were monitored by TLC and GC-MS of the reaction mixtures; after they were complete, simple work-up afforded the desired naphthofurans 16a and $\mathbf{1 6 b}$ and naphthothiophenes $\mathbf{1 6 c}$ and 16d. As in the benzothiophene preparations, the use of $\mathrm{ZnBr}_{2}$ for cyclization of naphthothio ketone $\mathbf{1 5 c}$ after 20 h of heating at $175-180^{\circ} \mathrm{C}$ did not give a substantial amount of $\mathbf{1 6 c}$, but led instead to the formation of di(1-naphthyl) disulfide ( $15 \%$ ) and 1,4-bis(4-chlorobenzoyl)-2,3-diphenylbutane (16\%). As in the case of $\mathbf{6 h}$, the last named product was a result of consecutive rearrangement-recombination of the radical formed after thermal decomposition of ketone 15c. Intermediates $\mathbf{1 3}$ and $\mathbf{1 5}$ could be isolated, as we demonstrated for the cases of 13a and $\mathbf{1 5 a}, \mathbf{1 5 b}$ and 15d, and then could be converted smoothly into the corresponding naphtho-furans and -thiophenes 16, as shown for the cases of 15a and 15d (see Experimental), which increases the synthetic utility of the new process.

## Conclusion

One-carbon homologation reaction of benzotriazole-containing (thio)alkoxy derivatives 4, $\mathbf{1 3}$ afforded a new series of $\alpha$-(thio)alkoxy-substituted ketones $\mathbf{6 a - h}$ and $\mathbf{1 5 a} \mathbf{- d}$, which were subsequently cyclized to give a wide range of various substituted benzo $[b]$ furans $\mathbf{7 a}-\mathbf{f}$ and benzo $[b]$ thiophenes $7 \mathbf{g}, \mathbf{h}$. Several compounds, fused benzo derivatives, were also prepared, i.e. naphtho[1,2-b]furans $\mathbf{1 6 a , b}$ and naphtho[1,2-b]thiophenes 16c,d. The overall yields of compounds $7 \mathbf{a}-\mathbf{h}$ were in the range of $49-80 \%$, while $\mathbf{1 6 a - d}$ yields were $71-83 \%$. The reaction sequence was performed in a one-pot manner, starting from the corresponding (benzotriazol-1-yl)methyl aryl (thio)ethers 3a-e and 12a,b. The procedure is simple, efficient, and more general than most of the previously used methods, thus adding a valuable alternative to the known reactions for the preparation of benzo[ $b]$ furans and benzo[b]thiophenes.

## Experimental

## General

Melting points were determined with a hot-stage apparatus and are uncorrected. NMR spectra were taken in $\mathrm{CDCl}_{3}$ with tetramethylsilane as the internal standard for ${ }^{1} \mathrm{H}(300 \mathrm{MHz})$ or solvent as the internal standard for ${ }^{13} \mathrm{C}(75 \mathrm{MHz}) . J$ Values are given in Hz. Tetrahydrofuran (THF) was distilled under nitrogen immediately prior to use from sodium-benzophenone. All reactions with air-sensitive compounds were carried out under an argon atmosphere. Column chromatography was conducted with silica gel 230-400 mesh.

General procedure for the preparation of 1-(aryloxymethyl)benzotriazoles 3a-c, 1-(arylthiomethyl)benzotriazoles 3d and 3e, 1-(1-naphthyloxymethyl)benzotriazole 12a, and 1-(1-naphthylthiomethyl)benzotriazole 12b
To a stirred solution of the corresponding (thio)phenol $\mathbf{2 a -} \mathbf{e}$ or (thio)naphthol 11a,b ( 15 mmol ) in ethanol $\left(50 \mathrm{~cm}^{3}\right), \mathrm{NaOH}$ $(15 \mathrm{mmol}, 0.60 \mathrm{~g})$ was added at room temperature. After 1 h , ethanol was evaporated in vacuo, and a solution of 1-chloromethylbenzotriazole $1(15 \mathrm{mmol}, 2.51 \mathrm{~g})$ in DMF $\left(50 \mathrm{~cm}^{3}\right)$ was added to the resulting solid. The reaction mixture was heated at
$65-70{ }^{\circ} \mathrm{C}$ for 15 h . On cooling, the mixture was poured into icewater to give crystals, which were filtered, washed with water $\left(100 \mathrm{~cm}^{3}\right)$ and dried to afford the pure products $\mathbf{3 a}-\mathbf{e}$ and $\mathbf{1 2 a}, \mathbf{b}$.

1-(Phenoxymethyl)benzotriazole 3a. Isolated as a white solid (3.24 g, $96 \%$ ), mp 64-65 ${ }^{\circ} \mathrm{C}$ (lit., ${ }^{16} 64{ }^{\circ} \mathrm{C}$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $6.56(2 \mathrm{H}, \mathrm{s}), 7.03(1 \mathrm{H}, \mathrm{t}, J 7.2), 7.09(2 \mathrm{H}, \mathrm{d}, J 8.2), 7.28(2 \mathrm{H}$, d, $J 7.2$ ), $7.41(1 \mathrm{H}, \mathrm{t}, J 8.0), 7.53(1 \mathrm{H}, \mathrm{t}, J 8.0), 7.70(1 \mathrm{H}, \mathrm{d}, J$ $8.3), 8.08(1 \mathrm{H}, \mathrm{d}, J 8.31) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 75.0,109.9$, $116.4,120.1,123.1,124.4,128.1,129.7,132.8,146.3$ and 156.2.

1-(2-Ethylphenoxymethyl)benzotriazole 3b. Isolated as a white solid ( $2.88 \mathrm{~g}, 76 \%$ ), mp $54-55^{\circ} \mathrm{C}$ (Found: N, 16.76. Calc. for $\left.\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}: \mathrm{N}, 16.60 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.10(3 \mathrm{H}, \mathrm{t}, J$ 7.4), $2.54(2 \mathrm{H}, \mathrm{q}, J 7.4), 6.59(2 \mathrm{H}, \mathrm{s}), 6.96-7.01(1 \mathrm{H}, \mathrm{m}), 7.14$ $7.19(3 \mathrm{H}, \mathrm{m}), 7.42(1 \mathrm{H}, \mathrm{t}, J 7.5), 7.54(1 \mathrm{H}, \mathrm{t}, J 7.5), 7.66(1 \mathrm{H}$, d, $J 8.2$ ), $8.09(1 \mathrm{H}, \mathrm{d}, J 8.2) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 14.2,22.9$, $75.0,109.7,114.0,120.0,123.0,124.3,127.0,128.0,129.5$, $132.8,133.9,146.2$ and 154.0.

1-(3,4,5-Trimethylphenoxymethylbenzotriazole 3c. Isolated as a white solid ( $3.65 \mathrm{~g}, 91 \%$ ), mp 98-99 ${ }^{\circ} \mathrm{C}$ (Found: C, 71.86; H, 6.75; N, 16.10. Calc. for $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}: \mathrm{C}, 71.99 ; \mathrm{H}, 6.41$; N , $15.72 \%) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.07(3 \mathrm{H}, \mathrm{s}), 2.21(6 \mathrm{H}, \mathrm{s}), 6.50$ $(2 \mathrm{H}, \mathrm{s}), 6.73(2 \mathrm{H}, \mathrm{s}), 7.40(1 \mathrm{H}, \mathrm{t}, J 7.5), 7.53(1 \mathrm{H}, \mathrm{t}, J 7.5)$, $7.70(1 \mathrm{H}, \mathrm{d}, J 8.2), 8.07(1 \mathrm{H}, \mathrm{d}, J 8.2) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $14.6,20.7,75.3,110.0,115.4,120.1,124.4,128.0,129.8,132.9$, 137.9, 146.4 and 153.6.

1-[(4-Chlorophenyl)thiomethyl]benzotriazole 3d. Isolated as a white solid ( $4.00 \mathrm{~g}, 97 \%$ ), mp $105-106^{\circ} \mathrm{C}$ (Found: C, 56.72; H, 3.57; N, 15.33. Calc. for $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{3} \mathrm{ClS}: \mathrm{C}, 56.72 ; \mathrm{H}, 3.66 ; \mathrm{N}$, $15.27 \%) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 5.93(2 \mathrm{H}, \mathrm{s}), 7.16-7.23(4 \mathrm{H}, \mathrm{m})$, $7.38-7.51(3 \mathrm{H}, \mathrm{m}), 8.07(1 \mathrm{H}, \mathrm{d}, J 8.2) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $52.6,110.0,120.3,124.3,127.6,129.5,130.3,132.1,134.4,135.3$ and 146.5 .

1-[(4-Methylphenyl)thiomethyl]benzotriazole 3e. Isolated as a white solid ( $3.56 \mathrm{~g}, 93 \%$ ), mp 104-105 ${ }^{\circ} \mathrm{C}$ (Found: N, 16.91. Calc. for $\left.\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{~S}: \mathrm{N}, 16.47 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.30(3 \mathrm{H}, \mathrm{s})$, $5.89(2 \mathrm{H}, \mathrm{s}), 7.03(2 \mathrm{H}, \mathrm{d}, J 8.0), 7.12(2 \mathrm{H}, \mathrm{d}, J 8.0), 7.35-7.46$ (3 $\mathrm{H}, \mathrm{m}), 8.08(1 \mathrm{H}, \mathrm{d}, J 8.3) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 21.1,53.2,110.2$, $120.1,124.1,127.3,128.2,130.1,132.2,133.5,139.1$ and 146.4.

1-(1-Naphthyloxymethyl)benzotriazole 12a. Isolated as a white solid ( $3.92 \mathrm{~g}, 95 \%$ ), mp 102-103 ${ }^{\circ} \mathrm{C}$ (Found: C, 74.11; H, 4.76; $\mathrm{N}, 15.36$. Calc. for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{O}: \mathrm{C}, 74.17 ; \mathrm{H}, 4.76$; N, $15.26 \%$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 6.65(2 \mathrm{H}, \mathrm{s}), 7.16(1 \mathrm{H}, \mathrm{d}, J 7.5), 7.31$ (2 H, t, J8.0), $7.37-7.48(4 \mathrm{H}, \mathrm{m}), 7.63(1 \mathrm{H}, \mathrm{d}, J 8.0), 7.74(1 \mathrm{H}, \mathrm{d}$, $J 7.4), 8.03(1 \mathrm{H}, \mathrm{d}, J 8.2), 8.11(1 \mathrm{H}, \mathrm{d}, J 8.5) ; \delta_{\mathrm{C}}(75 \mathrm{MHz} ;$ $\left.\mathrm{CDCl}_{3}\right) 74.8,108.0,109.6,119.9,121.3,122.6,124.3,125.4$, $125.6,125.7,126.5,127.5,128.1,132.7,134.5,146.2$ and 151.7.

1-(1-Naphthylthiomethyl)benzotriazole 12b. Isolated as white crystals ( $4.15 \mathrm{~g}, 95 \%$ ), mp $103-104^{\circ} \mathrm{C}$ (Found: C, 69.76; H, 4.57; $\mathrm{N}, 14.29$. Calc. for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~N}_{3} \mathrm{~S}: \mathrm{C}, 70.08 ; \mathrm{H}, 4.50 ; \mathrm{N}$, $14.42 \%) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 6.01(2 \mathrm{H}, \mathrm{s}), 7.30-7.46(6 \mathrm{H}, \mathrm{m})$, $7.61-7.78(4 \mathrm{H}, \mathrm{m}), 8.02(1 \mathrm{H}, \mathrm{d}, J 7.6) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $52.6,110.1,120.1,124.1,126.7,126.8,127.4,127.6,127.7$, $129.1,129.2,129.3,132.2,132.4,132.9,133.5$ and 146.4.

General procedure for the preparation of 1-(benzotriazol-1-yl)-1phenoxy 1-substituted methanes 4a,b, 1-(benzotriazol-1-yl)-1-(4methylphenylthio)ethane 4 h , and 1-(1-naphthyloxy)-1-
(benzotriazol-1-yl)-4-phenylbutane 13a
To a stirred solution of the corresponding 1-(aryloxymethyl)benzotriazoles 3a,b, 1-[(4-methylphenyl)thiomethyl]benzotriazole 3e, or 1-(1-naphthyloxymethyl)benzotriazole 12a (5 mmol ) in dry THF $\left(50 \mathrm{~cm}^{3}\right)$ at $-78^{\circ} \mathrm{C}$ under $\operatorname{argon} \mathrm{Bu}{ }^{n} \mathrm{Li}$ $\left(1.6 \mathrm{M}, 3.8 \mathrm{~cm}^{3}, 5.5 \mathrm{mmol}\right)$ was added. After 1 h , the appropriate electrophile ( 5.5 mmol ) [benzyl bromide $(0.94 \mathrm{~g})$ for $\mathbf{4 a}, n$-octyl iodide ( 1.32 g ) for $\mathbf{4 b}$, methyl iodide ( 0.78 g ) for $\mathbf{4 h}$, or 1-phenyl-3-bromopropane $(1.00 \mathrm{~g})$ for 13a] in THF $\left(5 \mathrm{~cm}^{3}\right)$ was added. The mixture was stirred at $-78^{\circ} \mathrm{C}$ for an additional 3 h and then at room temperature overnight. After being quenched with water $\left(50 \mathrm{~cm}^{3}\right)$, the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 50$ $\mathrm{cm}^{3}$ ) and the combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. The
solvent was evaporated in vacuo and the residue purified either by recrystallization or by column chromatography to give the corresponding pure product $\mathbf{4 a}, \mathbf{4 b}, 4 h$ or $\mathbf{1 3 a}$.

1-(Benzotriazol-1-yl)-1-phenoxy-2-phenylethane 4a. Purified by recrystallization from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$-hexane to give a white solid ( $1.50 \mathrm{~g}, 95 \%$ ), mp 89-90 ${ }^{\circ} \mathrm{C}$ (Found: C, 76.17; H, 5.47; N, 13.56. Calc. for $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}: \mathrm{C}, 76.17 ; \mathrm{H}, 5.43$; $\left.\mathrm{N}, 13.32 \%\right)$; $\delta_{\mathrm{H}}(300$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 3.55(1 \mathrm{H}, \mathrm{dd}, J 6.0,14.0), 3.79(1 \mathrm{H}, \mathrm{dd}, J 6.0$, 14.0), 6.85-7.00 ( $4 \mathrm{H}, \mathrm{m}$ ), 7.09-7.17 (3 H, m), 7.19-7.23 ( 2 H , $\mathrm{m}), 7.32(1 \mathrm{H}, \mathrm{t}, J 8.0), 7.43(1 \mathrm{H}, \mathrm{t}, J 8.0), 7.76(1 \mathrm{H}, \mathrm{d}, J 8.3)$, $8.01(1 \mathrm{H}, \mathrm{d}, J 8.3) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 41.4,89.0,111.0,116.4$, $120.2,123.1,124.2,127.4,127.7,128.6,129.4,129.6,131.7$, 134.5, 146.6 and 156.1.

1-(Benzotriazol-1-yl)-1-phenoxynonane 4b. Purified by column chromatography (hexane- $\mathrm{EtOAc}=8: 1$ ) to give a colourless oil ( $1.48 \mathrm{~g}, 88 \%$ ) (Found: C, 74.98; H, 8.46; N, 12.41. Calc. for $\left.\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}: \mathrm{C}, 74.74 ; \mathrm{H}, 8.06 ; \mathrm{N}, 12.45 \%\right) ; \delta_{\mathrm{H}}(300 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 0.87(3 \mathrm{H}, \mathrm{t}, J 6.5), 1.13-1.61(12 \mathrm{H}, \mathrm{m}), 2.25-2.40(1 \mathrm{H}$, m), 2.43-2.58 (1 H, m), 6.84-6.93 ( $2 \mathrm{H}, \mathrm{m}$ ), 6.98 ( $2 \mathrm{H}, \mathrm{d}, J 7.9$ ), $7.15(2 \mathrm{H}, \mathrm{t}, J 7.9), 7.30(1 \mathrm{H}, \mathrm{t}, J 7.6), 7.42(1 \mathrm{H}, \mathrm{t}, J 7.6), 7.82$ ( $1 \mathrm{H}, \mathrm{d}, J 8.4$ ), $8.02(1 \mathrm{H}, \mathrm{d}, J 8.4) ; \delta_{\mathrm{c}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.8$, $22.4,24.5,28.7,28.8,29.0,31.5,34.6,88.2,111.0,116.1,119.9$, 122.7, 124.0, 127.4, 129.4, 131.0, 146.6 and 156.1.

1-(Benzotriazol-1-yl)-1-(4-methylphenylthio)ethane 4h. Purified by column chromatography (hexane-EtOAc $=8: 1$ ) to give a colourless oil ( $1.21 \mathrm{~g}, 90 \%$ ) (Found: C, 66.75; H, 5.98; N, 15.74. Calc. for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{~S}: \mathrm{C}, 66.89$; $\mathrm{H}, 5.62$; $\mathrm{N}, 15.61 \%$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.09(3 \mathrm{H}, \mathrm{d}, J 7.1), 2.22(3 \mathrm{H}, \mathrm{s}), 6.25(1$ H, q, J7.1), 6.88-6.95 (4 H, m), $7.35(1 \mathrm{H}, \mathrm{t}, J 7.3), 7.44(1 \mathrm{H}, \mathrm{t}$, $J 7.3), 7.68(1 \mathrm{H}, \mathrm{d}, J 8.3), 8.03(1 \mathrm{H}, \mathrm{d}, J 8.3) ; \delta_{\mathrm{C}}(75 \mathrm{MHz} ;$ $\mathrm{CDCl}_{3}$ ) $20.6,21.0,63.1,111.1,120.0,123.8,126.9,127.2,129.7$, 131.4, 134.2, 139.2 and 146.5.

1-(1-Naphthyloxy)-1-(benzotriazol-1-yl)-4-phenylbutane 13a. Purified by column chromatography (hexane- $\mathrm{EtOAc}=8: 1$ ) to give a colourless oil ( $1.75 \mathrm{~g}, 89 \%$ ); $m / z(\mathrm{FAB}) 394.1913(\mathrm{M}+1)$. $\mathrm{C}_{26} \mathrm{H}_{23} \mathrm{~N}_{3} \mathrm{O}$ requires 394.1919; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 1.60-1.75$ $(1 \mathrm{H}, \mathrm{m}), 1.88-2.03(1 \mathrm{H}, \mathrm{m}), 2.42-2.54(1 \mathrm{H}, \mathrm{m}), 2.60-2.79$ (3 $\mathrm{H}, \mathrm{m}), 6.91(1 \mathrm{H}, \mathrm{d}, J 7.7), 7.03-7.52(12 \mathrm{H}, \mathrm{m}), 7.72(2 \mathrm{H}, \mathrm{d}, J$ $8.0), 8.01(1 \mathrm{H}, \mathrm{d}, J 8.2), 8.31(1 \mathrm{H}, \mathrm{d}, J 8.0)$ ) $\delta_{\mathrm{C}}(75 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) $26.2,34.2,34.9,87.7,107.7,111.0,120.1,121.3,122.4$, 124.3, 125.5, 125.6, 125.7, 126.0, 126.5, 127.7, 127.8, 128.3, 128.4, 131.1, 134.5, 141.0, 146.6 and 151.6.

## Preparation of 1-(4-methylphenyl)-2-(benzotriazol-1-yl)-2phenoxydecanol 5b

To a stirred solution of $\mathbf{4 b}(5 \mathrm{mmol})$ in dry THF $\left(50 \mathrm{~cm}^{3}\right)$ at $-78{ }^{\circ} \mathrm{C}$ under argon $\mathrm{Bu}^{n} \mathrm{Li}\left(1.6 \mathrm{~m}, 3.8 \mathrm{~cm}^{3}, 5.5 \mathrm{mmol}\right)$ was added. After 2 min , 4-methylbenzaldehyde ( $5.5 \mathrm{mmol}, 0.66 \mathrm{~g}$ ) in THF $\left(5 \mathrm{~cm}^{3}\right)$ was added. The mixture was stirred at $-78^{\circ} \mathrm{C}$ for an additional 3 h and then at room temperature overnight. After being quenched with water $\left(50 \mathrm{~cm}^{3}\right)$, the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}\left(3 \times 50 \mathrm{~cm}^{3}\right)$ and the combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. The solvent was evaporated in vacuo and the residue purified by column chromatography (hexane$\mathrm{EtOAc}=8: 1$ ) to give product $\mathbf{5 b}$ as a mixture of diastereomers; a colourless oil ( 1.26 g , isolated yield $55 \%$ ) (Found: N, 9.39. Calc. for $\mathrm{C}_{29} \mathrm{H}_{35} \mathrm{~N}_{3} \mathrm{O}_{2}: \mathrm{N}, 9.18 \%$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) (data in square brackets are given for minor isomer, the ratio of the products, according to the GC-MS data, is $2: 1) 0.85(3 \mathrm{H}, \mathrm{t}$, $J 5.3$ ), $0.98-1.26(10 \mathrm{H}, \mathrm{m}), 1.42-1.54(2 \mathrm{H}, \mathrm{m}), 2.26(3 \mathrm{H}, \mathrm{s})$ [2.20 (3 H, s)], 2.62-2.69 ( $2 \mathrm{H}, \mathrm{m}$ ), 3.62 ( $1 \mathrm{H}, \mathrm{br} \mathrm{s})$, [3.42 ( 1 H , br s)], $5.57(1 \mathrm{H}, \mathrm{s}),[5.43(1 \mathrm{H}, \mathrm{s})], 6.51-6.55(2 \mathrm{H}, \mathrm{m}), 6.72$ ( $2 \mathrm{H}, \mathrm{d}, J 8.1$ ) [6.65 ( $2 \mathrm{H}, \mathrm{d}, J 8.1$ )], [6.84 ( $2 \mathrm{H}, \mathrm{d}, J 7.8$ )], 6.91-6.97 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.02-7.11 ( $3 \mathrm{H}, \mathrm{m}$ ), 7.27-7.30 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.87-7.90 ( $1 \mathrm{H}, \mathrm{m}$ ), $7.97-8.03(1 \mathrm{H}, \mathrm{m})$; $\delta_{\mathrm{C}}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 14.0, 21.1 [20.9], 22.5 [22.4], 22.9, 28.8, 28.9, 29.4, 31.6, 32.8, 78.1 [77.3], 98.98 [99.0], 114.5 [113.3], 119.1 [119.2], 119.4 [119.3], 123.2 [123.4], 123.9 [123.7], 127.1 [127.06], 127.4, 128.6 [128.4], 129.4 [129.3], 134.1 [134.0], 134.2, 138.3 [138.2], 146.0 [145.9] and 153.9 [153.8].

General procedure for the preparation of $\alpha$-aryl(thio)oxy ketones 6a-e, 6 h and $\alpha$-naphthyl(thio) oxy ketones 15a, 15b and 15d To a stirred solution of the corresponding 1-(aryloxymethyl)benzotriazoles 3a,c, 1-[(4-methylphenyl)thiomethyl]benzotriazole 3e, 1-(1-naphthyloxymethyl)benzotriazole 12a, or 1-(1-naphthylthiomethyl)benzotriazole 12b ( 5 mmol ) in dry THF ( $50 \mathrm{~cm}^{3}$ ) at $-78^{\circ} \mathrm{C}$ under argon, $\mathrm{Bu}{ }^{n} \mathrm{Li}\left(1.6 \mathrm{~m}, 3.8 \mathrm{~cm}^{3}\right.$, 5.5 mmol ) was added. After 1 h , the appropriate electrophile $(5.5 \mathrm{mmol})$ [benzyl bromide $(0.94 \mathrm{~g})$ for $\mathbf{6 a}, n$-octyl iodide ( 1.32 g) for $\mathbf{6 b}$, methyl iodide ( 0.78 g ) for $\mathbf{6 c}$ and $\mathbf{6 h}, n$-butyl bromide $(0.75 \mathrm{~g})$ for $\mathbf{6 d}$ and $\mathbf{1 5 b}$, 1-phenyl-3-bromopropane ( 1.00 g ) for 15a, or $n$-propyl iodide ( 0.93 g ) for $\mathbf{6 e}$ and 15 d ] in THF $\left(5 \mathrm{~cm}^{3}\right)$ was added. The mixture was stirred at $-78^{\circ} \mathrm{C}$ for an additional 3 h and then at room temperature overnight. The solution formed was cooled again to $-78^{\circ} \mathrm{C}$, and another equivalent of $\mathrm{Bu}^{n} \mathrm{Li}\left(1.6 \mathrm{~m}, 3.8 \mathrm{~cm}^{3}\right.$, 5.5 mmol$)$ was added. After 2 min , the appropriate aldehyde ( 5.5 mmol ) [hydrocinnamaldehyde ( 0.74 g) for $\mathbf{6 a}, \mathbf{1 5 b}$ and $\mathbf{1 5 d}$, 4-methylbenzaldehyde $(0.66 \mathrm{~g})$ for $\mathbf{6 b}, \mathbf{6 c}$ and $\mathbf{6 h}, 2,4$-dichlorobenzaldehyde ( 0.96 g ) for $\mathbf{6 d}$, benzaldehyde $(0.58 \mathrm{~g})$ for 6 e , or 4-chlorobenzaldehyde $(0.77 \mathrm{~g})$ for 15 a$]$, in THF ( $5 \mathrm{~cm}^{3}$ ) was added. The mixture was stirred at $-78^{\circ} \mathrm{C}$ for an additional 3 h and then at room temperature overnight. A solution of $\mathrm{ZnBr}_{2}(12 \mathrm{mmol}, 2.7 \mathrm{~g})$ in THF $\left(10 \mathrm{~cm}^{3}\right)$ was added, the solvent was distilled off, and the oily residue was heated at $140^{\circ} \mathrm{C}$ for 10 h . Upon cooling, crude product was dissolved in $\mathrm{Et}_{2} \mathrm{O}$, filtered, quenched with water $\left(50 \mathrm{~cm}^{3}\right)$ and extracted with $\mathrm{Et}_{2} \mathrm{O}\left(3 \times 50 \mathrm{~cm}^{3}\right)$. The combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. The solvent was evaporated in vacuo and the residue purified by column chromatography to give the corresponding pure products 6 or 15 .

1,5-Diphenyl-2-phenoxypentan-3-one 6a. Purified by column chromatography (hexane-EtOAc $=20: 1$ ) to give a colourless oil ( $1.02 \mathrm{~g}, 62 \%$ ) (Found: C, $83.61 ; \mathrm{H}, 6.64$. Calc. for $\mathrm{C}_{23} \mathrm{H}_{22} \mathrm{O}_{2}$ : C, $83.60 ; \mathrm{H}, 6.71 \%) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.66-2.82(4 \mathrm{H}, \mathrm{m})$, $3.09(2 \mathrm{H}, \mathrm{d}, J 6.2), 4.73(1 \mathrm{H}, \mathrm{t}, J 6.2), 6.76(2 \mathrm{H}, \mathrm{d}, J 8.2), 6.94$ $(1 \mathrm{H}, \mathrm{t}, J 7.5), 7.05(2 \mathrm{H}, \mathrm{d}, J 7.7), 7.13-7.31(10 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}(75$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) 28.9, 38.4, 39.9, 83.7, 115.0, 121.6, 126.0, 126.8, 128.3, 129.5, 129.7, 136.2, 140.8157 .6 and 210.7.

1-(4-Methylphenyl)-2-phenoxydecan-1-one 6b. Purified by column chromatography (hexane- $\mathrm{EtOAc}=40: 1$ ) to give a colourless oil ( $1.28 \mathrm{~g}, 76 \%$ ) (Found: C, 81.31; H, 9.21. Calc. for $\left.\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{O}_{2}: \mathrm{C}, 81.60 ; \mathrm{H}, 8.94 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $0.87(3 \mathrm{H}, \mathrm{t}, J 4.7), 1.20-1.41(10 \mathrm{H}, \mathrm{m}), 1.42-1.71(2 \mathrm{H}, \mathrm{m})$, $1.84-2.21(2 \mathrm{H}, \mathrm{m}), 2.36(3 \mathrm{H}, \mathrm{s}), 5.26(1 \mathrm{H}, \mathrm{dd}, J 4.7,8.1), 6.82-$ $6.91(3 \mathrm{H}, \mathrm{m}), 7.14-7.25(4 \mathrm{H}, \mathrm{m}), 7.99(2 \mathrm{H}, \mathrm{d}, J 8.3) ; \delta_{\mathrm{C}}(75$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) 14.0, 21.5, 22.5, 25.7, 29.1, 29.2, 29.3, 31.7, 33.4, 81.3, 115.1, 121.1, 128.8, 129.3, 129.4, 131.9, 144.3, 157.9 and 198.5.

1-(4-Methylphenyl)-2-phenoxypropan-1-one 6c. Purified by column chromatography (hexane- $\mathrm{EtOAc}=8: 1$ ) to give white crystals ( $1.00 \mathrm{~g}, 84 \%$ ), mp $74-75^{\circ} \mathrm{C}$ (Found: C, 80.21; H, 6.96. Calc. for $\left.\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{O}_{2}: \mathrm{C}, 79.97 ; \mathrm{H}, 6.71 \%\right)$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) $1.68(3 \mathrm{H}, \mathrm{d}, J 6.8), 2.37(3 \mathrm{H}, \mathrm{s}), 5.45(1 \mathrm{H}, \mathrm{q}, J 6.8), 6.83-6.93$ $(3 \mathrm{H}, \mathrm{m}), 7.17-7.26(4 \mathrm{H}, \mathrm{m}), 7.98(2 \mathrm{H}, \mathrm{d}, J 8.3) ; \delta_{\mathrm{c}}(75 \mathrm{MHz} ;$ $\mathrm{CDCl}_{3}$ ) 18.7, 21.6, 76.5, 115.1, 121.3, 128.9, 129.4, 129.5, 131.6, 144.5, 157.4 and 198.4.

1-(2,4-Dichlorophenyl)-2-phenoxyhexan-1-one 6d. Purified by column chromatography (hexane- $\mathrm{EtOAc}=8: 1$ ) to give a colourless oil ( $1.10 \mathrm{~g}, 66 \%$ ) (Found: C, 63.89; H, 5.55. Calc. for $\left.\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{Cl}_{2} \mathrm{O}_{2}: \mathrm{C}, 64.11 ; \mathrm{H}, 5.38 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $0.89(3 \mathrm{H}, \mathrm{t}, J 7.2), 1.25-1.42(2 \mathrm{H}, \mathrm{m}), 1.45-1.61(2 \mathrm{H}, \mathrm{m})$, 1.92-1.99 ( $2 \mathrm{H}, \mathrm{m}$ ), $5.28(1 \mathrm{H}, \mathrm{t}, J 6.2), 6.86(2 \mathrm{H}, \mathrm{d}, J 8.0)$, $6.93(1 \mathrm{H}, \mathrm{t}, J 7.3), 7.19-7.27(3 \mathrm{H}, \mathrm{m}), 7.39(1 \mathrm{H}, \mathrm{s}), 7.47$ $(1 \mathrm{H}, \mathrm{d}, J 8.4) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.7,22.3,27.5,31.8,82.3$, $115.4,121.6,126.9,129.5,130.2,130.3,132.3,137.4,157.8$ and 200.5.

1-Phenyl-2-(2-ethylphenoxy)pentan-1-one 6e. Purified by column chromatography (hexane- $\mathrm{EtOAc}=40: 1$ ) to give white crystals ( $0.95 \mathrm{~g}, 72 \%$ ), mp $56-57^{\circ} \mathrm{C}$ (Found: C, 80.58 ; H, 8.02 . Calc. for $\left.\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{O}_{2}: \mathrm{C}, 80.82 ; \mathrm{H}, 7.85 \%\right)$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ )
$0.99(3 \mathrm{H}, \mathrm{t}, J 7.4), 1.25(3 \mathrm{H}, \mathrm{t}, J 7.5), 1.57-1.67(2 \mathrm{H}, \mathrm{m}), 1.98-$ $2.12(2 \mathrm{H}, \mathrm{m}), 2.75(2 \mathrm{H}, \mathrm{q}, J 7.1), 5.30(1 \mathrm{H}, \mathrm{dd}, J 4.7,8.3), 6.60$ $(1 \mathrm{H}, \mathrm{d}, J 8.0), 6.84(1 \mathrm{H}, \mathrm{t}, J 7.3), 7.00(1 \mathrm{H}, \mathrm{t}, J 7.3), 7.13(1 \mathrm{H}$, d, $J 7.2$ ), $7.43(2 \mathrm{H}, \mathrm{t}, J 7.6), 7.54(1 \mathrm{H}, \mathrm{t}, J 7.4), 8.09(2 \mathrm{H}, \mathrm{d}, J$ 7.2); $\delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) 13.8, 14.2, 19.1, 23.3, 35.4, 80.8, 111.4, 121.0, 126.6, 128.6, 128.7, 129.2, 132.8, 133.4, 134.5, 155.4 and 199.3.

1-(4-Methylphenyl)-2-(4-methylphenylthio)propan-1-one $6 \mathbf{h}$. Purified by column chromatography (hexane-EtOAc $=20: 1$ ) to give a colourless oil ( $1.09 \mathrm{~g}, 81 \%$ ) (Found: C, 75.32; H, 6.90 Calc. for $\left.\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{OS}: \mathrm{C}, 75.52 ; \mathrm{H}, 6.71 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) $1.51(3 \mathrm{H}, \mathrm{d}, J 6.9), 2.35(3 \mathrm{H}, \mathrm{s}), 2.44(3 \mathrm{H}, \mathrm{s}), 4.56(1 \mathrm{H}, \mathrm{q}, J$ 6.8 ), $7.10(2 \mathrm{H}, \mathrm{d}, J 8.0), 7.24-7.29(4 \mathrm{H}, \mathrm{m}), 7.89(2 \mathrm{H}, \mathrm{d}, J 8.2)$; $\delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 16.9,21.2,21.6,46.2,128.1,128.8,129.3$, 129.7, 133.2, 135.1, 138.9, 143.8 and 195.9.

1-(4-Chlorophenyl)-2-(1-naphthyloxy)-5-phenylpentan-1-one 15a. Purified by column chromatography (hexane-EtOAc $=$ $20: 1$ ) to give a colourless oil ( $1.53 \mathrm{~g}, 74 \%$ ) (Found: C, 78.18; H, 5.94. Calc. for $\mathrm{C}_{27} \mathrm{H}_{23} \mathrm{O}_{2} \mathrm{Cl}$ : C, $78.23 ; \mathrm{H}, 5.60 \%$ ); $\delta_{\mathrm{H}}(300 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 1.90-2.35(4 \mathrm{H}, \mathrm{m}), 2.73(2 \mathrm{H}, \mathrm{t}, J 7.4), 5.33(1 \mathrm{H}, \mathrm{dd}, J$ 4.1, 8.2), 6.57 ( $1 \mathrm{H}, \mathrm{d}, J 7.8$ ), 7.16-7.51 ( $11 \mathrm{H}, \mathrm{m}$ ), 7.75-7.78 ( 1 $\mathrm{H}, \mathrm{m}), 8.03(2 \mathrm{H}, \mathrm{d}, J 8.7), 8.35-8.38(1 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}(75 \mathrm{MHz} ;$ $\mathrm{CDCl}_{3}$ ) $27.2,32.7,35.3,82.0,105.6,121.2,121.9,125.5,125.6$, 125.7, 126.0, 126.6, 127.6, 128.4, 129.0, 130.2, 132.5, 134.7, 140.1, 141.4, 153.3 and 198.1.

1-Phenyl-4-(1-naphthyloxy)octan-3-one 15b. Purified by column chromatography (hexane- $\mathrm{EtOAc}=20: 1$ ) to give a colourless oil ( $1.39 \mathrm{~g}, 80 \%$ ) (Found: C, 83.22; H, 7.76. Calc. for $\left.\mathrm{C}_{24} \mathrm{H}_{26} \mathrm{O}_{2}: \mathrm{C}, 83.20 ; \mathrm{H}, 7.56 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 0.89(3 \mathrm{H}$, $\mathrm{t}, J 7.1), 1.28-1.56(4 \mathrm{H}, \mathrm{m}), 1.75-1.87(1 \mathrm{H}, \mathrm{m}), 1.89-2.02(1 \mathrm{H}$, m), 2.67-3.05 (4 H, m), 4.66 (1 H, dd, J4.9, 8.2), $6.52(1 \mathrm{H}, \mathrm{d}, J$ 7.5 ), 7.05 ( $2 \mathrm{H}, \mathrm{d}, J 6.6$ ), $7.10-7.29(5 \mathrm{H}, \mathrm{m}), 7.43(1 \mathrm{H}, \mathrm{d}, J 8.1)$, 7.48-7.52 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.79-7.82 ( $1 \mathrm{H}, \mathrm{m}$ ), 8.32-8.35 ( $1 \mathrm{H}, \mathrm{m}$ ); $\delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.8,22.4,27.5,29.0,32.0,38.5,83.4$, 105.2, 121.1, 122.0, 125.5, 125.6, 125.7, 126.0, 126.6, 127.6, 128.4, 134.7, 140.8, 153.5 and 211.5.

1-Phenyl-4-(1-naphthylthio)heptan-3-one 15d. Purified by column chromatography (hexane- $\mathrm{EtOAc}=8: 1$ ) to give a semi crystalline solid ( $1.62 \mathrm{~g}, 93 \%$ ) (Found: C, 79.04; H, 7.00. Calc. for $\left.\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{OS}: \mathrm{C}, 79.27 ; \mathrm{H}, 6.94 \%\right)$ ) $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 0.90$ ( $3 \mathrm{H}, \mathrm{t}, J 7.3$ ), 1.30-1.56 ( $2 \mathrm{H}, \mathrm{m}$ ), 1.64-1.87 (2 H, m), 2.83-2.95 ( $4 \mathrm{H}, \mathrm{m}$ ), 3.71 ( $1 \mathrm{H}, \mathrm{t}, J 7.5$ ), $7.11-7.25(6 \mathrm{H}, \mathrm{m}), 7.33(1 \mathrm{H}, \mathrm{d}, J$ 8.5), 7.42-7.49 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.70-7.77 ( $4 \mathrm{H}, \mathrm{m}$ ); $\delta_{\mathrm{C}}(75 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 13.7,20.5,30.0,32.4,40.9,56.9,126.0,126.2,126.4$ 126.6, 127.5, 127.7, 128.4, 128.6, 129.3, 131.2, 132.5, 133.6, 140.9 and 206.4.

General procedure for the preparation of polysubstituted benzo[b]furans 7a-d and 2-(3-phenylpropyl)-3-(4-chlorophenyl)-naphtho[1,2-b]furan 16a. A mixture of the corresponding $\alpha$ aryloxy ketone 6 or 1-(4-chlorophenyl)-2-(1-naphthyloxy)-5-phenylpentan-1-one 15a ( 5 mmol ) and $\mathrm{ZnBr}_{2}(10 \mathrm{mmol}, 2.3 \mathrm{~g})$ was heated at $175-180^{\circ} \mathrm{C}$ for $10-24 \mathrm{~h}$. On cooling, the oily residue was dissolved in $\mathrm{Et}_{2} \mathrm{O}$, filtered, quenched with water $\left(50 \mathrm{~cm}^{3}\right)$ and extracted with $\mathrm{Et}_{2} \mathrm{O}\left(3 \times 50 \mathrm{~cm}^{3}\right)$. The combined organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and the solvent was evaporated in vacuo. The crude product was purified by column chromatography to give the corresponding products 7 or 16a.

2-Benzyl-3-(2-phenylethyl)benzo[b]furan 7a. Purified by column chromatography (hexane-EtOAc $=40: 1$ ) to give a colourless oil ( $1.25 \mathrm{~g}, 80 \%$ ) (Found: C, 88.93 ; H, 6.72. Calc. for $\left.\mathrm{C}_{23} \mathrm{H}_{20} \mathrm{O}: \mathrm{C}, 88.43 ; \mathrm{H}, 6.45 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) \{data in square brackets are given for the regioisomer, 2-(2-phenylethyl)-3-benzylbenzo[b]furan; the ratio of the products, according to the GC-MS data, was 13:1\} 2.90-3.06 ( $4 \mathrm{H}, \mathrm{m}$ ), [3.84 ( $2 \mathrm{H}, \mathrm{s}$ )], $3.88(2 \mathrm{H}, \mathrm{s}), 7.07-7.30(14 \mathrm{H}, \mathrm{m})$, [7.37-7.53 $(14 \mathrm{H}, \mathrm{m})] ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 26.1,32.5,36.0,111.0,114.6$, 119.0, 122.1, 123.4, 126.1, 126.5, 128.3, 128.4, 128.5, 137.8, 141.6, 152.7 and 154.3.

2-Octyl-3-(4-methylphenyl)benzo[b]furan 7b. Purified by column chromatography (hexane-EtOAc $=40: 1$ ) to give a
colourless oil ( $1.26 \mathrm{~g}, 79 \%$ ) (Found: C, 86.34; H, 8.76. Calc. for $\left.\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{O}: \mathrm{C}, 86.20 ; \mathrm{H}, 8.81 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 0.87$ ( $3 \mathrm{H}, \mathrm{t}, J 6.7$ ), 1.18-1.40 ( $10 \mathrm{H}, \mathrm{m}$ ), 1.72-1.81 ( $2 \mathrm{H}, \mathrm{m}$ ), 2.43 $(3 \mathrm{H}, \mathrm{s}), 2.84(2 \mathrm{H}, \mathrm{t}, J 7.7), 7.18-7.30(4 \mathrm{H}, \mathrm{m}), 7.38(2 \mathrm{H}, \mathrm{d}$, $J 8.0)$, 7.45 ( $1 \mathrm{H}, \mathrm{d}, J 7.4$ ), $7.54(1 \mathrm{H}, \mathrm{d}, J 8.2) ; \delta_{\mathrm{C}}(75 \mathrm{MHz} ;$ $\mathrm{CDCl}_{3}$ ) 14.1, 21.3, 22.7, 26.8, 28.4, 29.2, 29.3, 29.4, 31.9, 110.8, 116.6, 119.5, 122.5, 123.4, 129.0, 129.1, 129.4, 129.9, 136.7, 154.0 and 155.2 .

2-Methyl-3-(4-methylphenyl)benzo[b]furan 7c. Purified by column chromatography (hexane- $\mathrm{EtOAc}=20: 1$ ) to give a colourless oil ( $0.88 \mathrm{~g}, 80 \%$ ); $m / z$ (EI) $222.1055\left(\mathrm{M}^{+}\right) . \mathrm{C}_{16} \mathrm{H}_{14} \mathrm{O}$ requires 222.1045 ; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.45(3 \mathrm{H}, \mathrm{s}), 2.55(3 \mathrm{H}$, s), $7.20-7.33(4 \mathrm{H}, \mathrm{m}), 7.41-7.48(3 \mathrm{H}, \mathrm{m}), 7.59(1 \mathrm{H}, \mathrm{d}, J 7.1)$; $\delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 12.8,21.2,110.7,116.8,119.4,122.5$, 123.4, 128.8, 129.4, 129.8, 130.0, 136.6, 151.0 and 154.1.

2-Butyl-3-(2,4-dichlorophenyl)benzo[b]furan 7d. Purified by column chromatography (hexane- $\mathrm{EtOAc}=40: 1$ ) to give a colourless oil ( $0.94 \mathrm{~g}, 59 \%$ ) (Found: C, 68.28 ; H, 5.54. Calc. for $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{Cl}_{2} \mathrm{O}: \mathrm{C}, 67.91 ; \mathrm{H}, 5.07 \%$ ); $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) 0.88 ( 3 $\mathrm{H}, \mathrm{t}, J 7.3), 1.28-1.40(2 \mathrm{H}, \mathrm{m}), 1.64-1.82(2 \mathrm{H}, \mathrm{m}), 2.58-2.80(2$ H, m), 7.18-7.37 ( $5 \mathrm{H}, \mathrm{m}$ ), $7.49(1 \mathrm{H}, \mathrm{d}, J 8.0)$, $7.58(1 \mathrm{H}, \mathrm{s})$; $\delta_{\mathrm{c}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.7,22.3,26.8,29.9,110.9,113.8,119.7$, 122.6, 123.6, 127.2, 128.8, 129.8, 130.4, 133.0, 134.3, 135.4, 153.9 and 156.4.

2-(3-Phenylpropyl)-3-(4-chlorophenyl)naphtho[1,2-b]furan
16a. Purified by column chromatography (hexane-EtOAc $=$ $40: 1$ ) to give a colourless oil ( $1.41 \mathrm{~g}, 71 \%$ ); $m / z$ (EI) 396.1298 $\left(\mathrm{M}^{+}\right) . \mathrm{C}_{27} \mathrm{H}_{21} \mathrm{ClO}$ requires $396.1281 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 2.13-$ $2.23(2 \mathrm{H}, \mathrm{m}), 2.71(2 \mathrm{H}, \mathrm{t}, J 7.4), 2.96(2 \mathrm{H}, \mathrm{t}, J 7.4), 7.14-7.29(6$ $\mathrm{H}, \mathrm{m}), 7.38-7.50(4 \mathrm{H}, \mathrm{m}), 7.57-7.68(3 \mathrm{H}, \mathrm{m}), 7.93(1 \mathrm{H}, \mathrm{d}, J 8.2)$, $8.32(1 \mathrm{H}, \mathrm{d}, J 8.2) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 26.3,30.1,35.3,117.4$, 118.1, 119.9, 121.2, 123.4, 123.7, 124.9, 125.9, 126.3, 128.4, $128.5,129.0,130.4,131.3,131.4,133.0,141.5,149.4$ and 154.2.

Preparation of 2-methyl-3-(4-methylphenyl)-5-methylbenzo[b]thiophene 7 h and 2-propyl-3-(2-phenylethyl)naphtho[1,2-b]thiophene 16d
To the corresponding 1-(4-methylphenyl)-2-(4-methylphenyl-thio)propan-1-one $6 \mathrm{~h}(5 \mathrm{mmol}, 1.28 \mathrm{~g})$ or 1-phenyl-4-(1-naphthylthio)heptan-3-one $\mathbf{1 5 d}(5 \mathrm{mmol}, 1.74 \mathrm{~g})$ polyphosphoric acid $(10 \mathrm{~g})$ was added at room temperature and the reaction mixture was heated at $120-125^{\circ} \mathrm{C}$ for 20 h . Upon cooling, the oily residue was dissolved in $\mathrm{Et}_{2} \mathrm{O}$, poured into ice-water $\left(100 \mathrm{~cm}^{3}\right)$ and extracted with $\mathrm{Et}_{2} \mathrm{O}\left(3 \times 50 \mathrm{~cm}^{3}\right)$. The combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was evaporated in vacuo. The residue was purified by column chromatography to give the corresponding product $\mathbf{7 h}$ or $\mathbf{1 6 d}$.

2-Methyl-3-(4-methylphenyl)-5-methylbenzo[b]thiophene 7h. Purified by column chromatography (hexane-EtOAc $=40: 1$ ) to give a colourless oil $(1.02 \mathrm{~g}, 81 \%)$ (Found: C, 80.98; H, 6.48 . Calc. for $\left.\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{~S}: \mathrm{C}, 80.91 ; \mathrm{H}, 6.39 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $2.36(3 \mathrm{H}, \mathrm{s}), 2.42(3 \mathrm{H}, \mathrm{s}), 2.45(3 \mathrm{H}, \mathrm{s}), 7.06-7.10(2 \mathrm{H}, \mathrm{m})$, 7.24-7.30 (4 H, m), $7.63(1 \mathrm{H}, \mathrm{d}, J 8.2) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $14.5,21.3,21.4,121.6,122.5,125.4,129.2,129.9,132.5,133.5$, 133.8, 135.3, 136.0 and 136.8.

2-Propyl-3-(2-phenylethyl)naphtho[1,2-b]thiophene 16d. Purified by column chromatography (hexane- $\mathrm{EtOAc}=40: 1$ ) to give white crystals ( $2.74 \mathrm{~g}, 83 \%$ ), mp $78-79^{\circ} \mathrm{C}$ (Found: C, 83.36; H, 6.53. Calc. for $\left.\mathrm{C}_{23} \mathrm{H}_{22} \mathrm{~S}: \mathrm{C}, 83.60 ; \mathrm{H}, 6.72 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $1.03(3 \mathrm{H}, \mathrm{t}, J 7.3), 1.67-1.79(2 \mathrm{H}, \mathrm{m}), 2.78(2 \mathrm{H}, \mathrm{t}, J 7.7), 3.09(2$ $\mathrm{H}, \mathrm{t}, J 8.1), 3.55(2 \mathrm{H}, \mathrm{t}, J 8.1), 7.27-7.39(5 \mathrm{H}, \mathrm{m}), 7.55(1 \mathrm{H}, \mathrm{t}$, $J 7.2), 7.65(1 \mathrm{H}, \mathrm{t}, J 8.3), 7.71(1 \mathrm{H}, \mathrm{d}, J 8.7), 7.84(1 \mathrm{H}, \mathrm{d}, J 8.7)$, $7.99(1 \mathrm{H}, \mathrm{d}, J 8.2), 8.72(1 \mathrm{H}, \mathrm{d}, J 8.3) ; \delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $14.0,24.9,30.6,31.1,35.8,120.9,123.2,124.5,126.0,126.2$, $128.5,129.3,129.9,132.2,133.2,133.8,136.7,141.0$ and 141.6.

One-pot procedure for the preparation of polysubstituted benzo[b]furans $7 \mathrm{a}-\mathrm{f}$, benzo[b]thiophenes 7 g , 7 h , naphtho[1,2-b]furans 16a, 16b, and naphtho[1,2-b]thiophenes 16c, 16d To a stirred solution of the corresponding 1-[aryloxy(thio)-
methyl]benzotriazoles 3a-3e or 1-[naphthyloxy(thio)methyl]benzotriazoles 12a,b ( 5 mmol ) in dry THF $\left(50 \mathrm{~cm}^{3}\right)$ at $-78^{\circ} \mathrm{C}$ under argon, $\mathrm{Bu}^{n} \mathrm{Li}\left(1.6 \mathrm{~m}, 3.8 \mathrm{~cm}^{3}\right.$, 5.5 mmol ) was added. After 1 h , the appropriate electrophile ( 5.5 mmol ) [benzyl bromide ( 0.94 g ) for $\mathbf{7 a}, \mathbf{g}$ and $\mathbf{1 6 c}, n$-octyl iodide ( 1.32 g ) for $\mathbf{7 b}$, methyl iodide ( 0.78 g ) for $7 \mathbf{c}, \mathbf{h}, n$-butyl bromide $(0.75 \mathrm{~g})$ for $7 \mathbf{d}$ and $\mathbf{1 6 b}, n$-propyl iodide ( 0.93 g ) for 7 e and $\mathbf{1 6 d}, n$-hexyl bromide ( 0.91 g ) for $7 \mathbf{f}$, and 1-bromo-3-phenylpropane ( 1.10 g ) for 16a] in THF $\left(5 \mathrm{~cm}^{3}\right)$ was added. The mixture was stirred at $-78^{\circ} \mathrm{C}$ for an additional 3 h and then at room temperature overnight. The reaction solution was cooled down to $-78{ }^{\circ} \mathrm{C}$ and a second equivalent of $\mathrm{Bu}^{n} \mathrm{Li}\left(1.6 \mathrm{~m}, 3.8 \mathrm{~cm}^{3}, 5.5 \mathrm{mmol}\right)$ was added. After 2 min , the appropriate aldehyde ( 5.5 mmol ) [hydrocinnamaldehyde ( 0.74 g ) for 7a,g and 16b,d, 4-methylbenzaldehyde $(0.66 \mathrm{~g})$ for $7 \mathbf{b}, \mathbf{c}, \mathrm{~h}, ~ 2,4$-dichlorobenzaldehyde $(0.96 \mathrm{~g})$ for 7 d , benzaldehyde $(0.58 \mathrm{~g})$ for 7 e , and 4 chlorobenzaldehyde ( 0.77 g ) for $\mathbf{7 f}$ and 16a,c] in THF $\left(5 \mathrm{~cm}^{3}\right)$ was added. The mixture was stirred at $-78^{\circ} \mathrm{C}$ for an additional 3 h and then at room temperature overnight. A solution of $\mathrm{ZnBr}_{2}(12 \mathrm{mmol}, 2.7 \mathrm{~g})$ in THF $\left(10 \mathrm{~cm}^{3}\right)$ was added, and the oily residue was heated at $140^{\circ} \mathrm{C}$ for 10 h .

## Method A: preparation of benzo[ $b$ ]furans 7a-f and naphtho-[1,2-b]furan 16a

The reaction mixture was heated at $175-180^{\circ} \mathrm{C}$ for $10-24 \mathrm{~h}$. Upon cooling, the oily residue was dissolved in $\mathrm{Et}_{2} \mathrm{O}$, filtered, quenched with water $\left(50 \mathrm{~cm}^{3}\right)$, and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 50$ $\left.\mathrm{cm}^{3}\right)$. The combined organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and solvent was evaporated in vacuo. The residue was purified by column chromatography to give the corresponding compounds 7a-f or 16a.

Method B: preparation of benzo[b]thiophenes 7 g , 7 h , naphtho-[1,2-b]furan 16b and naphtho $[1,2-b]$ thiophenes $16 \mathrm{c}, \mathrm{d}$
To the crude oily residue polyphosphoric acid ( 10 g ) was added at room temperature, and the reaction mixture was heated at $120-125^{\circ} \mathrm{C}$ for 20 h . Upon cooling, the oily residue was dissolved in $\mathrm{Et}_{2} \mathrm{O}$, poured into ice-water ( $100 \mathrm{~cm}^{3}$ ) and extracted with $\mathrm{Et}_{2} \mathrm{O}\left(3 \times 50 \mathrm{~cm}^{3}\right)$. The combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and the solvent was evaporated in vacuo. The crude product was purified by column chromatography to give the corresponding products $\mathbf{7 g}, \mathbf{7 h}$ or $\mathbf{1 6 b}-\mathbf{d}$. The appearance and NMR spectral data for compounds 7a-d, 7h and 16a, 16d were identical to those described above (cyclization of $\mathbf{6}$ into 7 and 15 into 16 ).
2-Propyl-3-phenyl-7-ethylbenzo[b]furan 7e. Purified by column chromatography (hexane-EtOAc $=40: 1$ ) to give a colourless oil ( $0.95 \mathrm{~g}, 72 \%$ ); $m / z$ (EI) $264.1511\left(\mathrm{M}^{+}\right) . \mathrm{C}_{19} \mathrm{H}_{20} \mathrm{O}$ requires 264.1514; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 0.99(3 \mathrm{H}, \mathrm{t}, J 7.3), 1.38(3 \mathrm{H}, \mathrm{t}$, $J 7.6), 1.75-1.87(2 \mathrm{H}, \mathrm{m}), 2.84(2 \mathrm{H}, \mathrm{t}, J 7.4), 2.96(2 \mathrm{H}, \mathrm{q}, J$ 7.6), 7.07-7.17 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.33-7.39 (2 H, m), 7.44-7.48 (4 H, m); $\delta_{\mathrm{c}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.9,14.2,21.9,23.0,28.8,117.0,122.7$, 122.8, 126.9, 127.4, 128.5, 128.7, 129.2, 133.2, 152.5 and 154.7.

2-Hexyl-3-(4-chlorophenyl)-4,5,6-trimethylbenzo[b]furan 7f. Purified by column chromatography (hexane-EtOAc $=40: 1$ ) to give a colourless oil ( $1.12 \mathrm{~g}, 64 \%$ ) (Found: C, 78.19 ; H, 8.08. Calc. for $\left.\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{ClO}: \mathrm{C}, 77.93 ; \mathrm{H}, 7.68 \%\right)$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) $0.87(3 \mathrm{H}, \mathrm{t}, J 6.0), 1.19-1.36(6 \mathrm{H}, \mathrm{m}), 1.61-1.70(2 \mathrm{H}, \mathrm{m}), 2.04$ $(3 \mathrm{H}, \mathrm{s}), 2.21(3 \mathrm{H}, \mathrm{s}), 2.41(3 \mathrm{H}, \mathrm{s}), 2.57(2 \mathrm{H}, \mathrm{t}, J 4.9), 7.17(1 \mathrm{H}$, s), $7.28(2 \mathrm{H}, \mathrm{d}, J 8.5), 7.41(2 \mathrm{H}, \mathrm{d}, J 7.8) ; \delta_{\mathrm{c}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $14.0,15.0,16.1,22.5,26.3,28.3,28.8,31.5,109.4,116.3,125.6$, $128.3,128.5,129.4,132.1,132.6,133.2,133.5,152.2$ and 154.9.

## 2-Benzyl-3-(2-phenylethyl)-5-chlorobenzo[b]thiophene $\quad \mathbf{7 g}$.

 Purified by column chromatography (hexane-EtOAc $=40: 1$ ) to give a colourless oil ( $1.41 \mathrm{~g}, 78 \%$ ) (Found: C, 76.50; H, 5.56. Calc. for $\left.\mathrm{C}_{23} \mathrm{H}_{19} \mathrm{ClS}: \mathrm{C}, 76.12 ; \mathrm{H}, 5.28 \%\right)$; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) $2.85(2 \mathrm{H}, \mathrm{t}, J 7.7), 3.09(2 \mathrm{H}, \mathrm{t}, J 7.7), 3.90(2 \mathrm{H}, \mathrm{s}), 7.09-7.16$ $(4 \mathrm{H}, \mathrm{m}), 7.21-7.32(7 \mathrm{H}, \mathrm{m}), 7.63-7.66(2 \mathrm{H} \mathrm{m}) ; \delta_{\mathrm{C}}(75$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $28.9,34.3,35.8,121.1,123.3,124.0,126.2,126.7$, $128.5,128.6,130.3,130.7,137.0,139.4,141.2,141.3$ and 141.5.2-Butyl-3-(2-phenylethyl)naphtho[1,2-b]furan 16b. Purified by column chromatography (hexane- $\mathrm{EtOAc}=40: 1$ ) to give a colourless oil ( $1.20 \mathrm{~g}, 73 \%$ ) (Found: C, 87.65; H, 7.74. Calc. for $\left.\mathrm{C}_{24} \mathrm{H}_{24} \mathrm{O}: \mathrm{C}, 87.76 ; \mathrm{H}, 7.36 \%\right) ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 0.92(3 \mathrm{H}$, t, $J 7.3$ ), 1.28-1.40 ( $2 \mathrm{H}, \mathrm{m}$ ), 1.58 ( 2 H , quintet, $J 7.4$ ), $2.61(2 \mathrm{H}$, $\mathrm{t}, J 7.4), 2.94-2.98(4 \mathrm{H}, \mathrm{m}), 7.13-7.28(5 \mathrm{H}, \mathrm{m}), 7.43(1 \mathrm{H}, \mathrm{t}, J$ 7.0), 7.52-7.63 (3 H, m), 7.91 ( $1 \mathrm{H}, \mathrm{d}, J 8.2$ ), 8.26 ( $1 \mathrm{H}, \mathrm{d}, J 8.2$ ); $\delta_{\mathrm{C}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 13.8,22.4,26.1,30.7,36.5,114.4,118.1$, 119.8, 121.3, 122.5, 124.4, 124.8, 125.96, 126.0, 128.3, 128.4, 128.6, 130.9, 141.7, 149.1 and 154.5.

2-Benzyl-3-(4-chlorophenyl)naphtho[1,2-b]thiophene 16c. Purified by column chromatography (hexane-EtOAc $=40: 1$ ) to give white crystals ( $1.49 \mathrm{~g}, 78 \%$ ), mp 123-124 ${ }^{\circ} \mathrm{C}$ (Found: C, $78.21 ; \mathrm{H}, 4.50$. Calc. for $\left.\mathrm{C}_{25} \mathrm{H}_{17} \mathrm{ClS}: \mathrm{C}, 78.11 ; \mathrm{H}, 4.46 \%\right)$; $\delta_{\mathrm{H}}(300$ MHz; $\mathrm{CDCl}_{3}$ ) $4.03(2 \mathrm{H}, \mathrm{s}), 7.14(2 \mathrm{H}, \mathrm{d}, J 6.9), 7.20-7.35(6 \mathrm{H}$, m), $7.40(1 \mathrm{H}, \mathrm{t}, J 7.4), 7.48-7.53(3 \mathrm{H}, \mathrm{m}), 7.69(1 \mathrm{H}, \mathrm{d}, J 8.8)$, $7.79(1 \mathrm{H}, \mathrm{d}, J 8.8), 7.88(1 \mathrm{H}, \mathrm{d}, J 8.0) ; \delta_{\mathrm{c}}\left(75 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $34.8,120.5,123.4,124.8,125.2,125.7,126.5,128.5,128.6$, 129.3, 129.6, 131.7, 132.0, 133.7, 133.9, 135.1, 136.7, 136.8, 140.0 and 141.2.

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